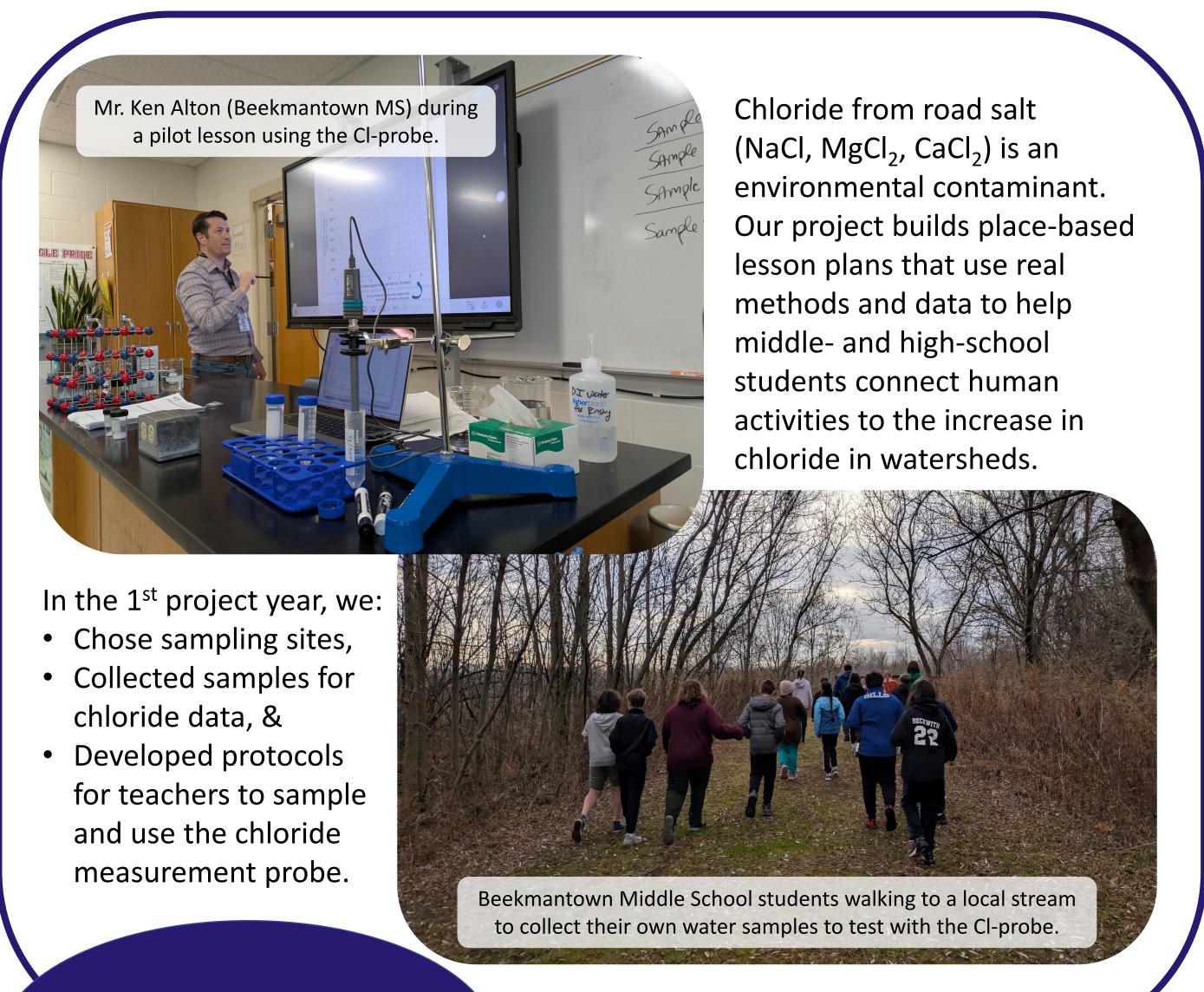
Sampling Salty Water in Champlain Basin Streams: Educational and Informational

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Purpose

We chose sites based on the following criteria:

- Proximity to a USGS gauging station.
- Distance from SUNY Plattsburgh campus.
- Water within reach from bridge or stream bank with a 16 ft (~4.8 m) telescoping pole.
- Bridge or bank is safe for sampling in all weather.
- Safe parking and walking route to the site.
- Permission to access via private property.



The location of possible sampling sites within the Lake Champlain Basin. The Little Chazy, Saranac, Little Ausable, and Ausable Rivers met our criteria for long-term sampling for this project.



Sharing the site-selection process with middle- and high-school students teaches them about sampling methods, including challenges and limitations. This helps them understand more about Planning & Carrying Out an Investigation, an important part of the state science standards.

Site Selection









STATE UNIVERSITY OF NEW YORK PLATTSBURGH

Sampling

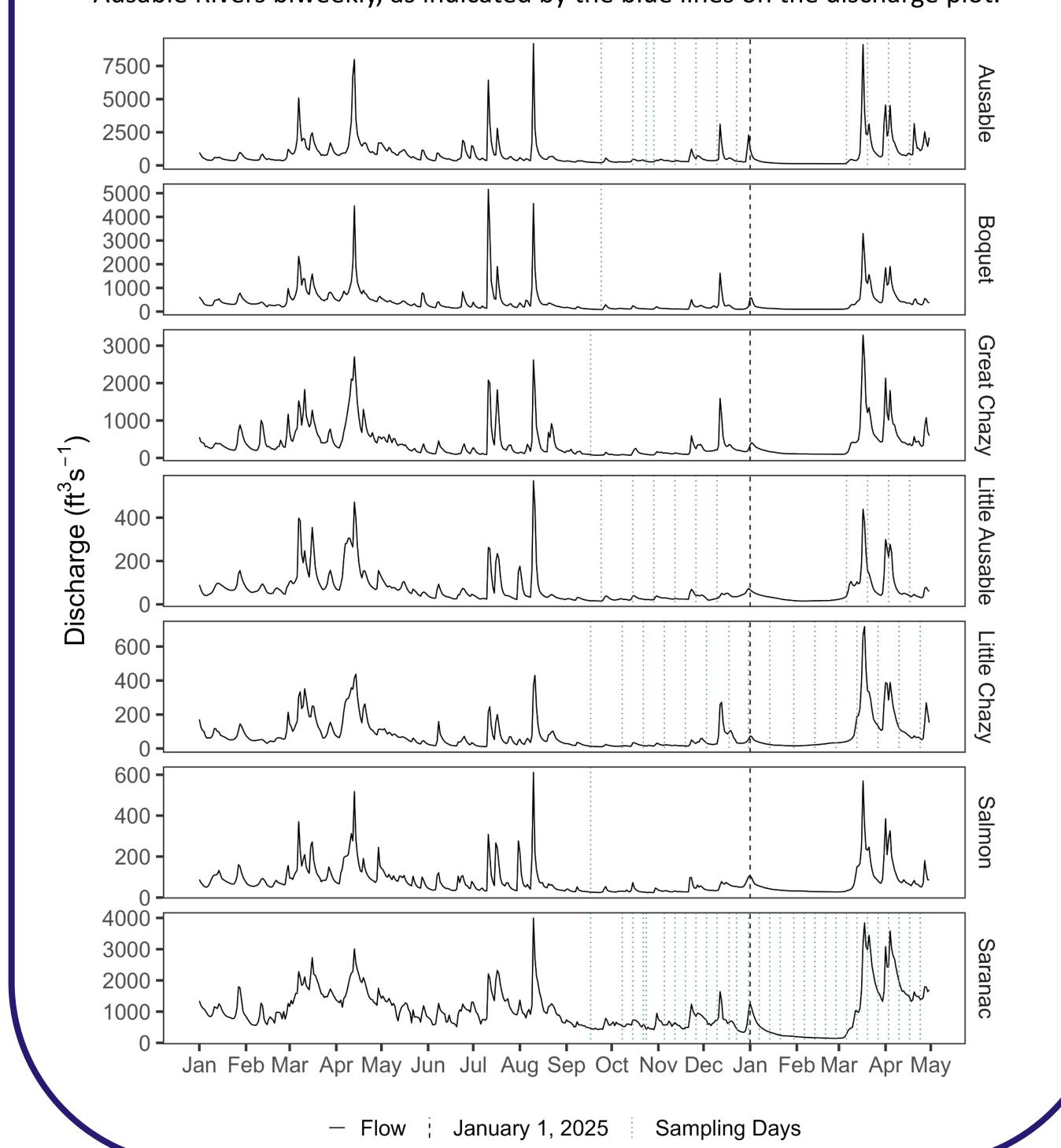
sampling poles for loan to schools, and we developed a sampling protocol for participating teachers. This protocol needed to be easy to follow, produce reliable chloride data, and work in all weather conditions.

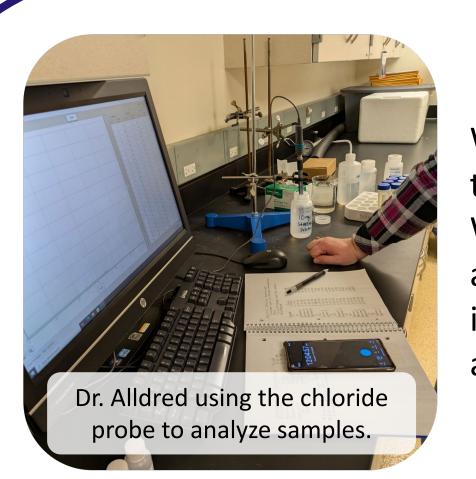
We made sampling kits with telescoping



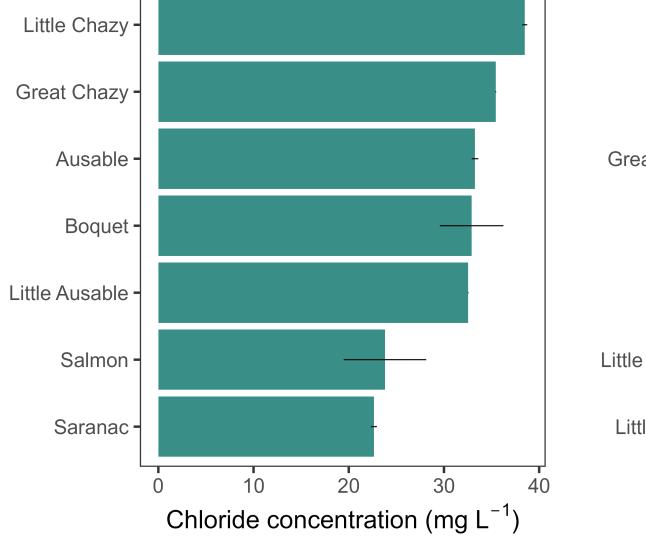


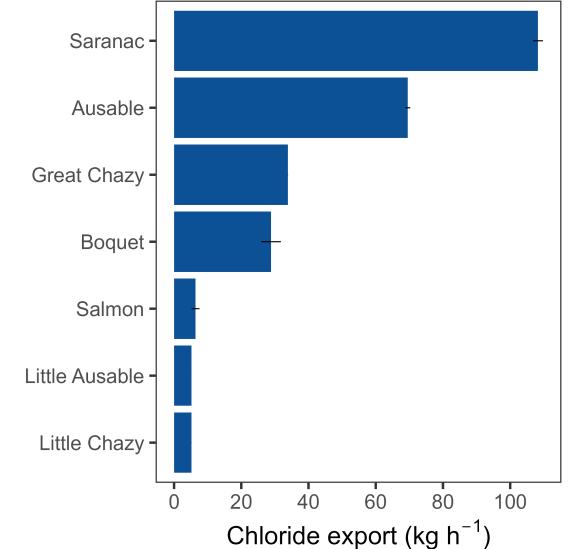
We sampled the Saranac River weekly and the Little Chazy, Little Ausable, and Ausable Rivers biweekly, as indicated by the blue lines on the discharge plot.



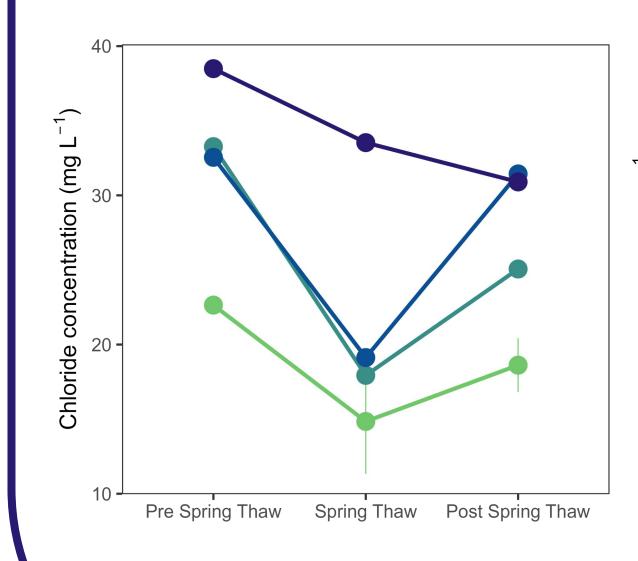


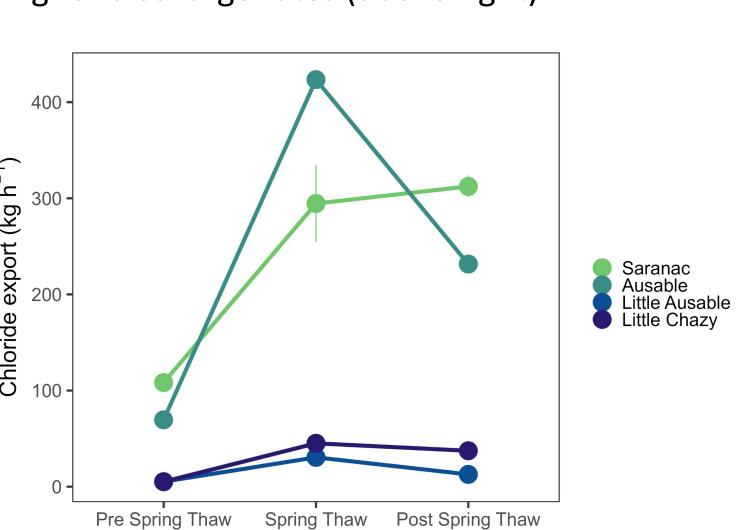
We included a Vernier® chloride probe in the sampling kits for loan to schools. We developed a user-friendly protocol for analyzing the chloride concentration (mg L⁻¹) in river samples that produced repeatable and reliable measurements.





Preliminary data showed that the Little Chazy River had the highest chloride concentration (above left), but the Saranac River exported the greatest amounts of chloride (kg h⁻¹) due to higher discharge rates (above right).





We were able to show a correlation between peaks in chloride exported and the timing of road salt use.

Results

Future

USING THE CHLORIDE ION PROBI We have three water-salt mixtures. We made these by adding regular table salt bought from the store to a small beaker of water. Table salt is a mixture of sodium ions (Na) and chlorine ions (Cl). Amount of Chlorine Ions (Cl) (milligrams of Cl per Liter) Use the amount of chlorine ions we measured in Sample 1 and Sample 2 to make a prediction for how much chlorine ions we will find in Sample 3 and Sample 4. Put this in the chart above. 5. Why did you choose the numbers of your prediction for Samples 1 & 2? What is your reasoning? A page from an activity we piloted at Beekmantown Middle School that used the Cl-probe.

The end goal of this project is to build middleand high-school lesson plans (with feedback from area teachers) that allow students to:

- Collect and analyze their own samples for elevated chloride using kits developed as part of the project, and
- 2. Query a database of chloride data that we collected to test hypotheses.

We will continue sampling for two years to provide data needed for these lesson plans.